

Names, Formulas, and Charges of Common Polyatomic Ions

Positive Ion (Cation)		
1+	Ammonium	NH_4^+
Negative Ions (Anions)		
1-	Acetate	$\text{C}_2\text{H}_3\text{O}_2^-$
	Bromate	BrO_3^-
	Chlorate	ClO_3^-
	Chlorite	ClO_2^-
	Cyanide	CN^-
	Hydride	H^-
	Hydrogen Carbonate (bicarbonate)	HCO_3^-
	Hydrogen Sulfate (bisulfate)	HSO_4^-
	Hydroxide	OH^-
	Hypochlorite	ClO^-
	Iodate	IO_3^-
	Nitrate	NO_3^-
	Nitrite	NO_2^-
	Perchlorate	ClO_4^-
	Permanganate	MnO_4^-
	Thiocyanate	SCN^-
2-	Carbonate	CO_3^{2-}
	Chromate	CrO_4^{2-}
	Dichromate	$\text{Cr}_2\text{O}_7^{2-}$
	Oxalate	$\text{C}_2\text{O}_4^{2-}$
	Peroxide	O_2^{2-}
	Sulfate	SO_4^{2-}
	Sulfite	SO_3^{2-}
3-	Arsenate	AsO_4^{3-}
	Phosphate	PO_4^{3-}
	Phosphite	PO_3^{3-}