Name:	_ Section:

Names, Formulas, and Charges of Common Polyatomic Ions

	Positive Ion (Cation)		
1+	Ammonium	NH ₄ ⁺	
	Negative Ions (Anions)		
1-	Acetate	$C_2H_3O_2^-$	
	Bromate	$\mathrm{BrO_3}^-$	
	Chlorate	ClO ₃ ⁻	
	Chlorite	$\mathrm{ClO_2}^-$	
	Cyanide	CN ⁻	
	Hydride	H^-	
	Hydrogen Carbonate (bicarbonate)	HCO_3^-	
	Hydrogen Sulfate (bisulfate)	$\mathrm{HSO_4}^-$	
	Hydroxide	OH^-	
	Hypochlorite	ClO-	
	Iodate	IO_3^-	
	Nitrate	NO_3^-	
	Nitrite	NO_2^-	
	Perchlorate	${ m ClO_4}^-$	
	Permanganate	$\mathrm{MnO_4}^-$	
	Thiocyanate	SCN ⁻	
2–	Carbonate	CO_3^{2-}	
	Chromate	CrO ₄ ²⁻	
	Dichromate	$Cr_2O_7^{2-}$	
	Oxalate	$C_2O_4^{2-}$	
	Peroxide	O_2^{2-}	
	Sulfate	SO_4^{2-}	
	Sulfite	SO ₃ ²⁻	
3–	Arsenate	AsO_4^{3-}	
	Phosphate	PO_4^{3-}	
	Phosphite	PO ₃ ³⁻	