Data and	Calculations	for Ex	periment 18	

Test Tube	Initial glucose concentration (from bottle)	Absorbance at 730 nm
1		
2		
3		
4		
5		
Unknown Code	To be determined in Question #1 below	

Based on your Excel<sup>®</sup> graph, what is the equation of the line?

## Questions

1. Using your unknown absorbance value and calibration line, what is the concentration of your unknown solution? Show your work below.

2. The following absorbance values for four solutions with known MnO<sub>4</sub><sup>-</sup> concentrations were measured using a spectrophotometer:

Solution	[MnO4 <sup>-</sup> ]	Absorbance
1	7.00 x 10 <sup>-5</sup> M	0.175
2	1.00 x 10 <sup>-4</sup> M	0.250
3	2.00 x 10 <sup>-4</sup> M	0.500
4	3.50 x 10 <sup>-4</sup> M	0.875

Plot a graph of Absorbance vs. Concentration of  $MnO_4^-$  using Microsoft Excel<sup>®</sup> (be sure to include your graph with this report). Determine the concentration of an unknown  $MnO_4^-$  sample whose absorbance is found to be 0.780.

3. In your own words, write a logical, coherent conclusion which demonstrates a thorough working knowledge and understanding of important concepts and underlying chemical principles pertinent to this experiment, forms appropriate conclusions based on interpretations of results, includes applications of and improvements in the experiment, and demonstrates accountability by providing justification for any errors. If additional space is needed, please use the back of this page. (For additional guidelines on writing this conclusion, please refer to the **Moorpark College Chemistry Department Laboratory Report Rubric** found in the lab manual and department website.)