

Name: _____

Section: _____

Use the SDS provided in lab to answer the following questions:

1. List other names that are synonyms of sodium hydroxide and its formula.
2. What is its melting point?
3. What is done in case of contact with eyes?
4. How should a small spill be handled?
5. What procedure should be done if the substance is swallowed?
6. What are the NFPA Ratings for Health? Fire? Reactivity? Specific Hazard?
7. List three chemicals that should not be stored with NaOH.
8. How should solid NaOH be properly stored?

Name: _____

Section: _____

Your instructor will assign you a specific chemical compound along with SDS, and you should fill-in the table below with as much information as possible (note: several areas will remain blank) using the various resources listed below:

Name of Substance _____

Chemical Formula _____

| | Reagent Bottle | SDS Sheet | Merck Index | CRC or Lange's |
|------------------------|----------------|-----------|-------------|----------------|
| Other Names | | | | |
| Formula Weight | | | | |
| State of matter | | | | |
| Melting point | | | | |
| Boiling point | | | | |
| Density | | | | |
| Percent Composition | | | | |
| Soluble solvents | | | | |
| Manufacturer | | | | |
| Chemical Properties | | | | |
| Toxicity | | | | |

Contrast the differences between the four reference materials used above, and be specific.